

## THE BARBIER REACTION AND RELATED ONE STEP PROCESSES

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Since the 1970s the use of metals, other than magnesium, for one-step synthetic processes has gained tremendous momentum.

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The subtitle of this monograph, Related One-Step Processes, may require the following clarification. A synthesis with the aid of an organometallic compound is usually carried out in two steps: 1) the organometallic reagent is prepared from an organo halide and a metal in a solvent, 2) a chosen substrate is added to this reagent. In a Barbier reaction these two steps are combined: a mixture of the organo halide and the substrate is added to a suspension of the metal in an appropriate solvent.

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The one-step organic synthesis with metals discovered by Barbier almost 100 years ago was widely applied with magnesium as the main metal until the end of the 1960s.

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